

Balancing Chemical Equations Worksheet

- $\text{Mg} + \text{O}_2 \rightarrow \text{MgO}$
- $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$
- $\text{MgBr}_2 + \text{Cl}_2 \rightarrow \text{MgCl}_2 + \text{Br}_2$
- $\text{Ca(OH)}_2 + \text{HNO}_3 \rightarrow \text{Ca(NO}_3)_2 + \text{H}_2\text{O}$
- $\text{Al} + \text{HCl} \rightarrow \text{AlCl}_3 + \text{H}_2$
- $\text{H}_2 + \text{N}_2 \rightarrow \text{NH}_3$
- $\text{AgNO}_3 + \text{NaCl} \rightarrow \text{AgCl} + \text{NaNO}_3$
- $\text{K} + \text{H}_2\text{O} \rightarrow \text{KOH} + \text{H}_2$
- $\text{Al(OH)}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + \text{H}_2\text{O}$
- $\text{NaI} + \text{Cl}_2 \rightarrow \text{NaCl} + \text{I}_2$
- $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
- $\text{HgO} \rightarrow \text{Hg} + \text{O}_2$
- $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$
- $\text{C}_8\text{H}_{18} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- $\text{CaCO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{CaSO}_4 + \text{H}_2\text{O} + \text{CO}_2$
- $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$
- $\text{Mg(OH)}_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$
- $\text{AgNO}_3 + \text{NaBr} \rightarrow \text{AgBr} + \text{NaNO}_3$
- $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$
- $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$
- $\text{Na} + \text{O}_2 \rightarrow \text{Na}_2\text{O}$
- $\text{P}_4 + \text{O}_2 \rightarrow \text{P}_4\text{O}_{10}$
- $\text{Fe} + \text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + \text{H}_2$
- $\text{C} + \text{H}_2 \rightarrow \text{CH}_4$
- $\text{Na}_2\text{SO}_4 + \text{CaCl}_2 \rightarrow \text{CaSO}_4 + \text{NaCl}$
- $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- $\text{Al}_2\text{O}_3 \rightarrow \text{Al} + \text{O}_2$