

Respiration & ATP

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Stage I Biology

Cell Chemistry

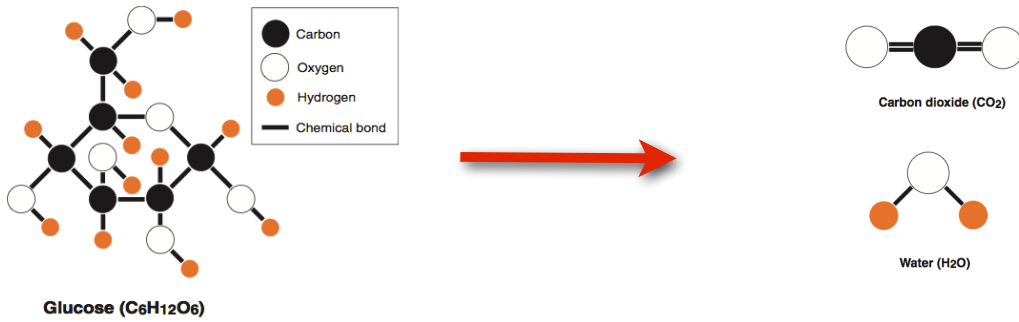
cell metabolism =

All metabolic reactions involve changes in energy (form or storage):

a) **Anabolic** Reactions (synthesis):

b) **Catabolic** Reactions (break down):

Energy in Bonds

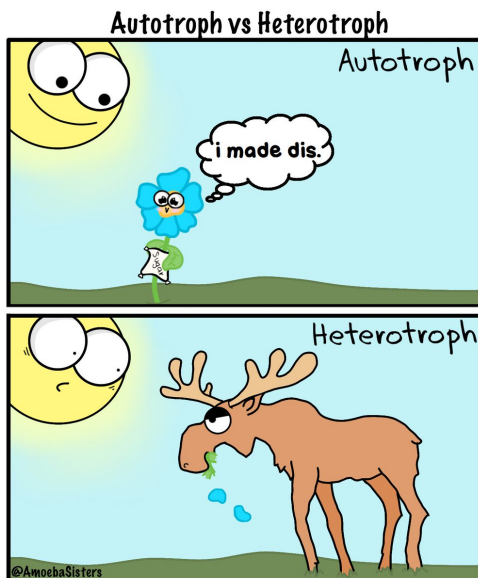


energy rich
reactants/substrates

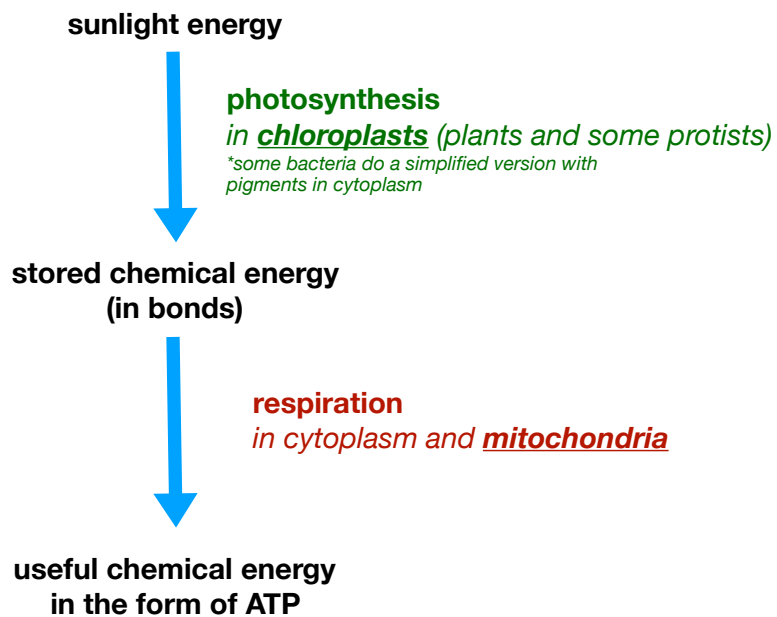
energy poor
products

Autotrophs vs Heterotrophs

Distinguish between autotrophs and heterotrophs.

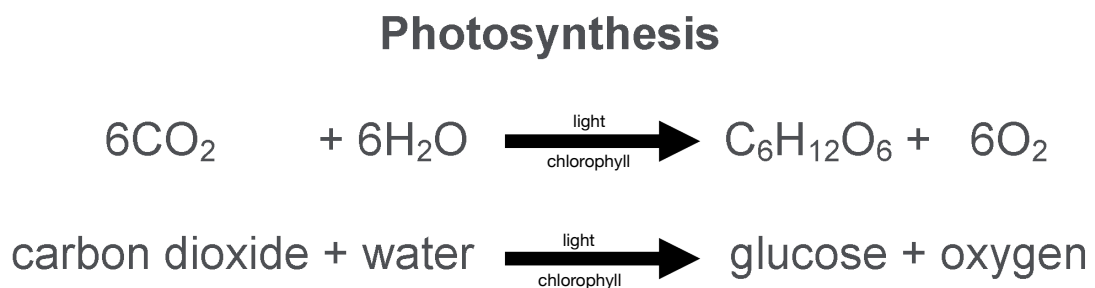


Sunlight for Photosynthesis



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Photosynthesis



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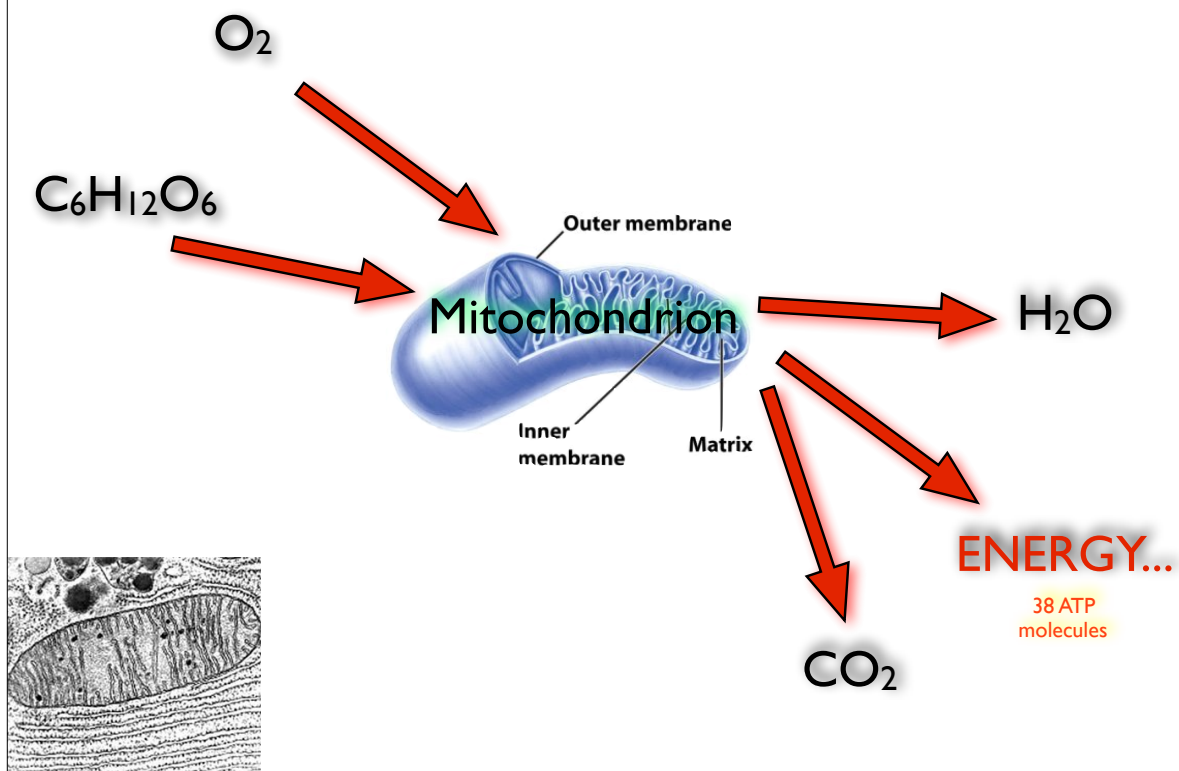
Aerobic Respiration

Glucose + Oxygen \longrightarrow Carbon Dioxide + Water + Energy



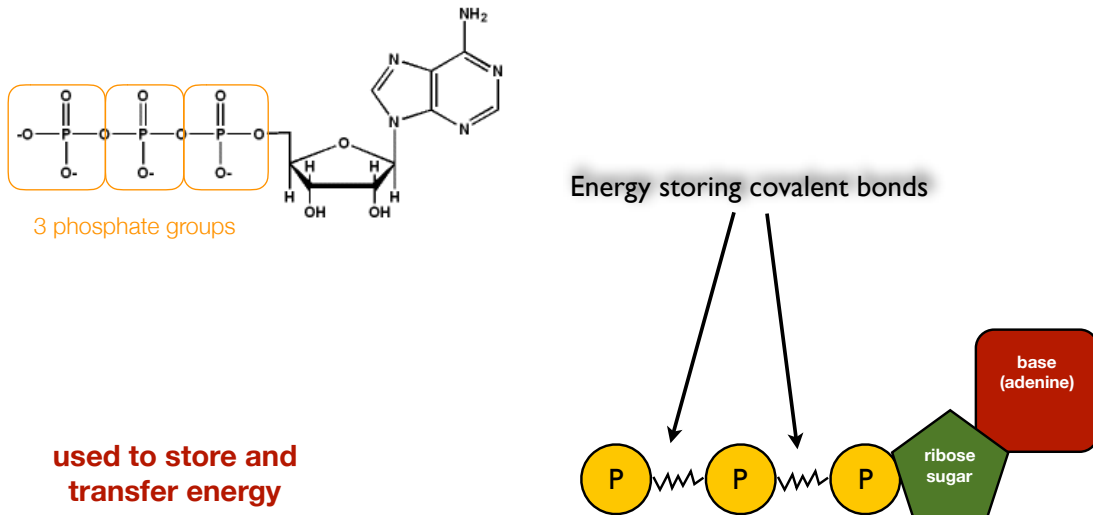
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Respiration!



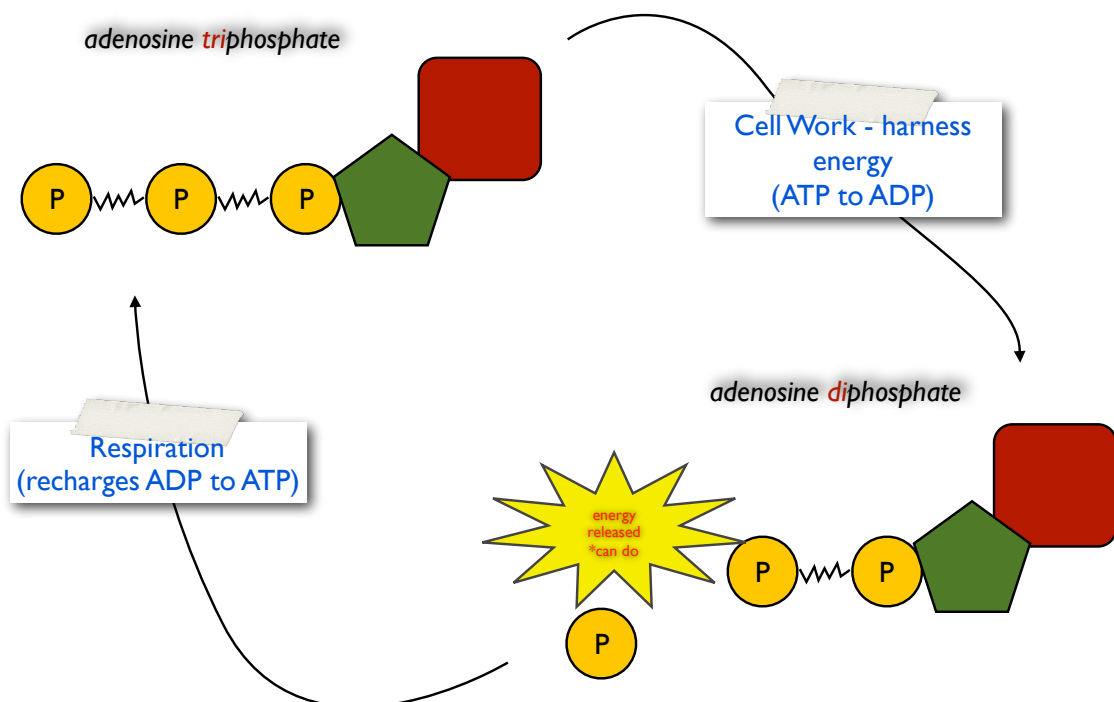
Formation of ATP

What is ATP?
adenosine *tr*iphosphate

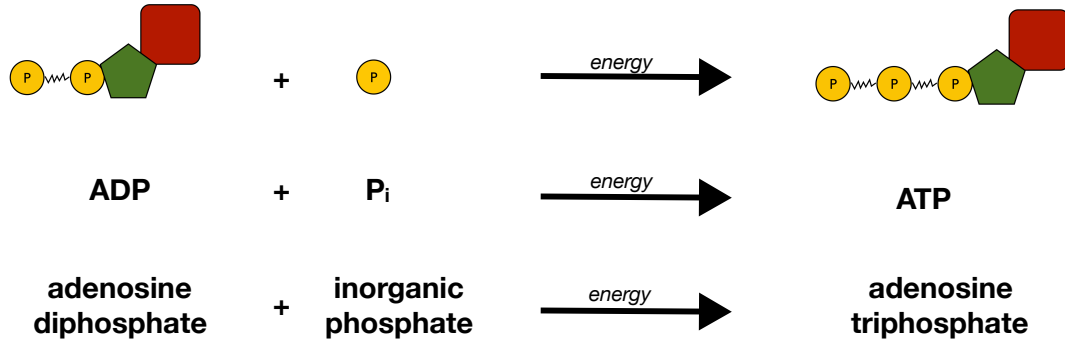


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How does it work?



Formation of ATP

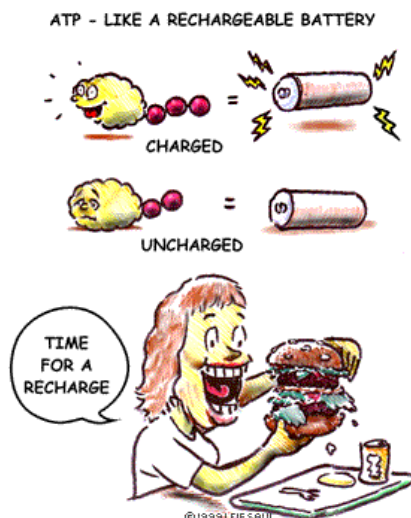


ADP and P_i are both negatively charged ions in the cytoplasm. Energy is required to overcome the repulsion to bond them.

This energy used to bond them is stored in the chemical bond between the two reactants.

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Formation of ATP

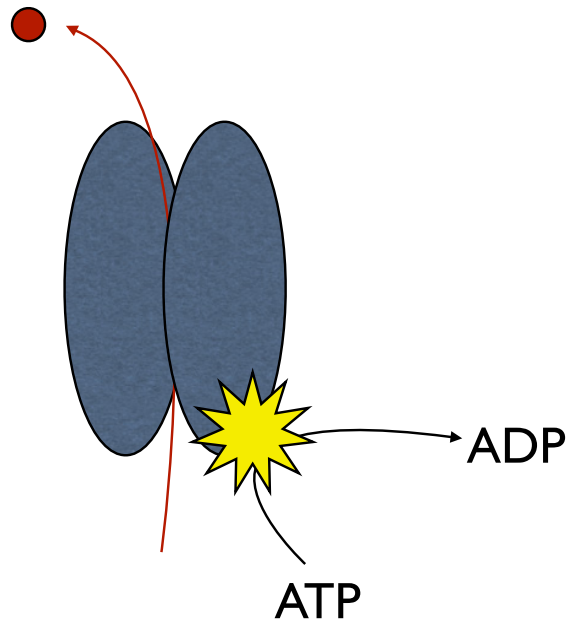


The energy used by human cells requires the hydrolysis of **100 to 150** moles of ATP daily, which is around 50 to 75 kg.

A human will typically use up his or her body weight of ATP over the course of the day. Each equivalent of ATP is recycled **500-750** times during a single day ($100 / 0.2 = 500$).

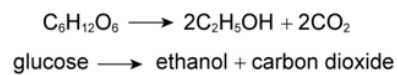
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Active Transport



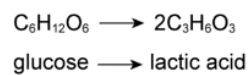
Anaerobic Respiration (fermentation)

In plants and yeast:



alcohol fermentation

In animals (and some bacteria):



lactic acid fermentation

Energy Release Efficiency (aerobic respiration vs fermentation)

Respiration produces energy in the usable form of ATP molecules.

Aerobic Respiration = 38 net ATP

Alcohol Fermentation = 2 net ATP

Lactic acid Fermentation = 2 net ATP